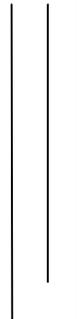


kj ffrn lj Zj lj Bfno

ufj7ufpf, dfj/È



lj 1fg tyf klj lw ; sfo

kj z k/liff tyf lj Bfyl{egf{; DaQWL lj lgodfj nL @)&&

# kj{rn ljZljBfno uf7ufpFdf7.

## .kj z k/Liff tyf laBfyl{egf{; DaGwl ljlgodfj nl @)&&U

kj{rn laZlaBfno lj 1fg tyf klj lw ; sfo cGtut{; #flnt ljleGg z]lfs sfoGmdx?sf]kj z k/Liff tyf laBfyl{egf{klqmf0{yk ; b0, kf/bzl{; dofgsh agfpg kj{rn laZlaBfno lj 1fg tyf klj lw ; sfosf] kj z k/Liff tyf laBfyl{egf{; DaGwl ljlgodfj nl @)&&U agf0{nfu'ug{f-5glo blvPsf]5 .

!= ; Hft gfd / kl/De

(s) of]ljlgodfj nlsf] gfd "kj{rn ljZljBfno lj 1fg tyf klj lw ; sfosf]kj z k/Liff tyf laBfyl{egf{; DaGwl ljlgodfj nl @)&&" /xg5 .

(v) of]ljlgodfj nl z]lfs aif{@)&&÷)\* ; fn blv nfu'xg5 .

@ kl/efiff: laifo j f k] un]csf]cy{gnfu0f o; ljlgodfj nldfM

!= "ljlgodfj nldfM" eGfn]kj{rn laZlaBfno lj 1fg tyf klj lw ; sfosf]kj z k/Liff tyf laBfyl{egf{; DaGwl ljlgodfj nl @)&& eGg]; Demgkb5 .

@ "kj z k/Liff" eGfn]kj{rn laZlaBfno lj 1fg tyf klj lw ; sfon]; #fng ug{ kj z k/Liff eGg]a%g' kb5 .

#= ":gfts" tx eGfn]kj{rn laZlaBfno lj 1fg tyf klj lw ; sfo cGtu{ xfn ; #flnt :gfts txsf sfoGmdx? eGg]a%gkb5 .

\$= ":gfts0/" tx eGfn]kj{rn ljZljBfno lj 1fg tyf klj lw ; sfo cGtu{ xfn ; #flnt :gfts0/ txsf sfoGmdx? eGg]a%gkb5 .

%= "csef/" eGfn]kj z k/Liffdf ; DalGwt lj ifox? Aff6 ; flwg] kZgx?sf] ; Wof eGg] a%gkb5 .

^= "kGM kj z k/Liff" eGfn]klxnfl]k6s kj z k/Liff e}; sk5L xg] bf}f] k/Liff eGg] a%gkb5 .

&= .8lg] eGfn]; sfosf]8lg eGg]a%gkb5 .

\*= ..; sfo] eGfn]kj{rn ljZljBfno cGtu{sf ; sfo eGg]; Demgkb5 .

(= 'kj z k/Liff ; ldt' eGfn]kj{rn ljZljBfno lj 1fg tyf klj lw ; sfosf]kj z k/Liff ; #fngsf]nflu u7g ul/Psf]; ldt eGg]a%gkb5 .

!)= .of]otf qmd] -Merit List\_ eGfn]kj z k/Liffdf k]kt k]kts cg' f/ ; aGbf a9L cs Nofpg]nf0{klxnfl]qmdf /fvl tof/ ul/Psf]laBfyl{?sf]List eGg]a%gkb5 .

!! +ljZljBfno] eGfn]kj{rn ljZljBfno eGg]a%gkb5 .

#= kj z k/Liffsf]km/fd tyf cfjZos sfuhft ; DaGwdf

(s\_ ; sfoaf6 ln0g]kj z k/Liffsf]; rgf /fli60 blgs klqs / laZlaBfnofsf]j]; f06df k\$flzt u/lg]5 .

(v\_ ; sfoaf6 ePsf]kj z k/Liffsf]; rgf cg' f/ laBfly{?n] cfknh] c]bbog ug{ rfx\$]f] Zflfs sfoGmd 5gf] ul/ ; DalGwt snh] af6 g}km/fd eg]kg5 .

(u\_ ; DkOf{SofDk; x?n]krf/fd egIcalw ; dfkt ePk5L kšlzt ; rgf adfllhdsf]; do l; df leq o; lalgodfj nl cg'; Ĥl @ df ePsf]; mUg 9fřdf excel sheet df laBfyl{x?sf]laj/0fsf]SofDk; kđV äf/f kđfłoft Hard copy tyf soft copy -CD/Pen Drive df /fvL\_ jf Software SofDk; sf] cflwsf/s kq, krf/d afkt hDdf u/\$f] /sdsf]; Ssn aš eř/, / krf/dx? SofDk; sf] cflwsf/ls kl|tlglw ; lxt 8lg sfořno jf kj ž k/Llff ; Dačwl ; rgfdf tfl]sPsf] :yfgdf aenřpg' kgš . ; dol; df leq lgod cg'; f/ krf/fdx? křkt ge0{; d:of ePdf ; Dalčwt SofDk; sf] SofDk; kđV lhDd]f/ xgš . ; fy}SofDk; n]k7fPsf]laBfyl{x?sf]la/a/0fdf km/s kl/ sg}qbl ePdf ; Dalčwt SofDk; lhDd]f/ xgkgš .

(3\_ kř ž kl/lffdf krf/d egIlaBfyl{x?sf]ofłotf ku]gkušf] cfaZos sfuhftx? b?:t eP gePsf] kOf{hfr ul/ ; Dalwt SofDk; kđVn]?h'u/\$f]xgkgš . ; Dalčwt SofDk; kđVn]?h'gu/\$f] kj ž kl/lff krf/d :jls[t ul/g]5č . ljZj lj Bfnon]třšř]dfkb08 tyf Ğčtd ofłotf cg'; f/ krf/d e/\$f]gePdf pQm krf/d :j tM vf/h -c:jls[\_ xgš / ; řš]lhDd]f/L :jod\; Dalčwt SofDk; ÷snh÷lj Bfyl{xgš .

(^\_ 5fqj [tsf]xsdf ; Dalčwt SofDk; n]5fqj [t lj godfj nl cg'; f/ krf/d e/f0{k7fpg' kgš .

### \$= laBfyl{egf{; Dačwl Joj :yf

(s\_ ; Dačwg křkt SofDk; x?n]sg}zřfs sfočmd nuftf/ # -tlg\_ aif{Dd ; řfng gu/\$f]v08df kj ž k/Llff cuřj}ljZj lj Bfnon] třšř] kgM ; řfngsf] kšřf křf ul/ dfq pQm zřfs sfočmddf egf{křpgš .

### %= kj ž k/Llff ; ldlit

k= lj= sfošf/L kl/ifbn]kj ž k/Llff ; řfng ugšf]nflu 8lg jf 8lgn]třšř]JolStsf]; ofřstj df kj ž k/Llff ; ldlit u7g ugš . kj ž k/Llff 8lg sfořnoaf6 ; řfng xgš . kj ž k/Llff ; ldltn]kj ž k/Llffn0{ Aojl:yt u/L ; řfng ug{cfaZos sfořwłx? Afřpg ; Sgš . ul7t ; ldltn]kj ž k/Llffsf]; DkOf{klqmf křf ul/ glthf kšřzg ugš . kj ž k/Llff ; ldltn]cfaZostf cg'; f/ lj1x? / ; Dalčwt 8lg sfořnosf sd{řf/lx? ; dř /xg]u/L ljlečg pk; ldlit u7g ul/ sfo{ug{; Sgš . lj1x? ř sd{řf/lx?nf0{o; ljlgodfjnlsf]cg'; Ĥl % df pNnř eP cg'; f/sf]kf/L>lds pknAw u/f0čš .

### ^= kř: kj ž k/Llff ; Dačwl Joj :yf

křf{rn ljZj lj Bfno lj 1fg tyf klj lw ; šfo čtuř ; řflnt :gfts tx Pj ÷gřtsřf/ txd f egfšf nflu ; řfng xg]kj ž k/Llff ; Dačwl lj 1fg tyf klj lw ; šfoaf6 tof/ u/L k= lj= lj 1fg tyf klj lw ; šfo lj Bf kl/ifbaf6 křl/t e0{cfPsf]"kj ž k/Llff ljlgodfj nl @)&&" sř]bkř g=^ čtuř křMkj ž k/Llff ; Dačwl Aoj :yf Mkyd kj ž k/Llff eP/ laBfyl{egf{e0 ; sš5l ljZj lj Bfnon]lgwf{of u/\$f]sřřf -zřfs sfočmd cg'; f/\_ !) kl|tzt jg ; ř]ečbf a9L egf{gePdf ; Dalčwt snřx?n]lnlv t ?kdf 8lg sfořnodf lgačg lbg' kgš . ; ř]kZřft ; DkOf{snřx?sf]hDdf sřřf / lallb fyl{x?sf]sh egf{; Wofnf0{ljZnřf of u/L !) kl|tzt jf ; ř]ečbf a9L egf{gePdf dfq 8lg sfořnoaf6 kgM kj ž k/Llff ; řfngsf]nflu křf{rn ljZj lj Bfno sfošf/L kl/ifbsf]! :js [L cg'; f/ břř]kj ž k/Llff ug{; Sgš . ; fy}kyd kj ž k/Llffsf]glthf kšřlzt e0{egfšf nflu lb0Psf]; dofj lw ; dfkt ePsf]@% lbg leqdf břř]kj ž k/Llff ; Dkčg ul/; Sg' kgš ečg]pNnř ul/P adfllhd o; nf0{křl/t ug]lgofč ul/of].

**&= kj ẏ k/lifsf]; do ldt tyf sᄁ ; Daᄁwdf**

(s\_ kj ẏ k/lifsf]sᄁ, ldt, ; do / :yfg kj ẏ k/liff ; ldt]sf]lg0fo{adfl]hd xgᄁ . ; f]; Daᄁwl hfgsf/l clud ?kdf ; dodfg}/fli6ᄁ blgs klqsf=lj Zj lj Bfnosf]j] ; f06df kᄁflzt ul/g]5 .  
 (v\_ lj Bfyl[?n]; Dalᄁwt Sofᄁk; af6 g)kj ẏ kq lngkgᄁ . lagf kj ẏ-kqsf]kj ẏ k/liff lbg kf0g] 5ᄁ .

**\*= kj ẏ k/liff glthf kᄁfzg ; Daᄁwl Aoj :yf**

(s) kj ẏ k/liffdf kᄁkt u/ᄁf]cᄁsf]cfwf/df glthf kᄁfzg ul/g]5 .

**( ẏkj ẏ k/liff ul/g]dlxgfm**

(s) :gfts / lklhl8l; Psf]txsf]xsdf >fj0f- ebf]-cgbflgt\_ dlxgdf ul/g]5 .  
 -v\_ :gftsᄁ/ txsf]xsdf dᄁ / -kᄁf -cgbflgt\_ dlxgdf ul/g]5 .

## cg' ᄁL !

**cg' ᄁL != lj 1fg tyf kljlw ; sfo kj ẏ k/liff ; ldt]sf]sfo]jlwl**

**cg' ᄁL != kj ẏ k/liff glthf kᄁfzg lawl**

(s) kj ẏ k/liff pᄁl0f{ug{:gfts txsf]xsdf ᄁᄁtd ##% cᄁ kᄁkt u/ᄁf]xgkgᄁ t/ Bachelor of Veterinary Science and Animal Husbandry sfoᄁᄁsf]xsdf sldtdf %) % cᄁ kᄁkt u/ᄁf]xgkgᄁ . :gftsᄁ/sf]xsdf ᄁᄁtd \$)% cᄁ kᄁkt u/ᄁf]xgkgᄁ . PGDCA sf]xsdf ᄁᄁtd ##% cᄁ kᄁkt u/ᄁf]xgkgᄁ  
 (v) laBfyl[?n]kj ẏ k/liffdf kᄁkt u/ᄁf sh kᄁkts+a/fa/ ePdf lghx?sf]! tx dlgsf] kᄁkts kᄁtztsf]cfwf/df ; ᄁL kᄁflzt ul/gᄁ .

**cg' ᄁL !=@ kj ẏ k/lifsf]cᄁef/ ; Daᄁwl Joj :yf**

(s) kj ẏ k/liffdf ; a}txdf a:tut (Objective) kᄁgx? !)) cᄁsf]; flwg]5 .

(v) :gfts txsf] xsdf lgᄁg cg' f/ cᄁ laefhg xgᄁ . Syallbus cg' ᄁL # df lb0Psf]5 .

; dk -s_		; dk -v_		; dk -u_	
lj ifo	c°ef/	lj ifo	c°ef/	lj ifo	c°ef/
English	!%	English	!%	English	@%
Math	#%	Math	!%	Math	@%
Physics	#)	Physics	!%	Basic Computer	@%
Chemistry	@)	Chemistry	@)	Appititude Test	@%
		Biology	#%		

(u\_ :gfts f/ txsf]xsf] lgDg cg' f/ c\$ laefhg xg\$ .

0lGhgo/l <sup>a</sup> tkm	
lj ifox?	c°ef/
Math	@)
Aptitude Test	!)
Related Subjects	&)

gg\0lGhgo/l <sup>a</sup> tkm	
lj ifox?	c°ef/
Aptitude Test	#)
Related Subjects	&)

(3\_ PGDCA sf]nflu Basic Computer, English Pj d\Basic Math laifox?j f6 kZgx? ; flwg]5 .  
c\$ laefhg lgDg cg' f/ xg\$ .

lj ifox?	c°ef/
Basic Computer	#)
English	#%
Basic Math	#%

(<sup>a</sup>\_ :gfts / :gfts f/ / PGDCA txsf]k] Z k/lifsf]; dofj wl @ 366sf]xg\$ .

**cg' P! != laBfyl{egf{: Da0wl Joj :yf**

(s\_ egf{xg] laBfyl{h] clgafo{?kdf k'f{rn lj Zj lj Bfno lj 1fg tyf klj lwsf] k] Z k/liff pQlOf{u/\$f]xg' kg\$ . :gfts txdf lj Bfyl{h] hg ; dxdf Kf] Z k/liff pQlOf u/\$f]xf]; flx ; dxdf dfq egf{lng ; Sg\$g . t/ ; dx s / ; dx v df ptlOf u/\$f]lj Bfyl{; dx u sf]z}lfs sfoqmdsf]dfkb08 cg' f/ egf{lng kfp]5g\ ; dxsf]lj efhg lgDgfg' f/ ul/Psf]5 .

**; dx SM**

BE (Civil, Electrical, Electronics & Communication, Computer, Mechanical, Geomatic) B. Arch. and Other Related Engineering Program

**; dx vM**

Bachelor in: Agriculture, Biomedical, Bio-Technology, B-Tech Biotechnology, Biochemistry, B. Tech. Food Technology, B. Tech. Dairy Technology, B. Sc. Forestry, Nutrition and other related program

**; dx uM**

BCA and BIT

**; dx 3M**

PGDCA

**cg' P! != k] Z k/liffdf ; lDdInt xg rflxg]Gogtd oflotf**

**:gfts txsf]nflu**

- **; dx -s\_ sf]nfluM**. Sc./10+2/A Level jf ; f]; /xsf]k/liffdf sh kOff{ssf]sldtdf Gogtd \$%% jf ul}Bssf] xsdf k]of\$ ljifodf GotQd c u\$ / CGPA 2.0 or more kft u/l ptlOf{ePsf]xgkg]5 . kltzt tyf u\$ bj }ePsf]xsdf kltztnf0{dfGotf lb0g\$ .

**Bachelor of Science (Honours) Agriculture Program**

1. The Candidate must have passed I. Sc. Examination or 10+2 (Science) with Physics, Chemistry, Biology & English from recognized Universities minimum 45% or GPA 2.00.
2. The Candidate must have passed entrance examination conducted by the University.
3. The Successful candidate in the entrance examination will be admitted in the merit basis in the University affiliate colleges.

**Bachelor's of Degree in Biomedical Engineering**

1. The Candidate must have passed I. Sc. Examination or 10+2 (Science) or Diploma in Engineering (Electrical and Electronics) from recognized Universities minimum 45% or GPA 2.00. The Candidate must have passed entrance examination conducted by the University.
2. The Successful candidate in the entrance examination will be admitted in the merit basis in the University affiliate colleges.

**Bachelor of Science in Biotechnology**

1. The Candidate must have passed PCL Examination or 10 + 2 (Biology Physical group) with minimum 45% or GPA 2.00 or Diploma in Engineering (Electrical and Electronics) from recognized Universities minimum 50% in aggregate in Examinations.
2. The Candidate must have passed entrance examination conducted by the University.
3. The Successful candidate in the entrance examination will be admitted in the merit basis in the University affiliate colleges.

**Bachelor of Technology in Biotechnology**

1. The Candidate must have passed PCL Examination or 10 + 2 (Biology or Physical group) with minimum 45% or GPA 2.00 or Diploma in Engineering (Electrical and Electronics) from recognized Universities minimum 50% in aggregate in Examinations.
2. The Candidate must have passed entrance examination conducted by the University.
3. The Successful candidate in the entrance examination will be admitted in the merit basis in the University affiliate colleges.

**Bachelor in Biochemistry**

1. The Candidate must have passed I. Sc. Examination or 10+2, of Science stream or equivalent examinations from recognized institution are eligible for admission. Students securing a minimum 45% or GPA 2.00.
2. The Candidate must have passed entrance examination conducted by the University.

### **Bachelor in Food & Dairy Technology**

1. The Candidate who have passed I. Sc., 10+2 (Science), Diploma in Food and Dairy Technology from an institution duly recognized by PU shall be eligible to apply for admission. Students securing a minimum 45% or GPA 2.00.
2. The Candidate must have passed entrance examination conducted by the University.

### **Bachelor of Science in Forestry**

1. The Candidate who have passed I. Sc. or 10+2 with biology or mathematics or Certificate forestry or 3 years diploma in forestry from recognized Universities minimum 50% marks in aggregate or C+ (CGPA 2.40)
2. The Candidate must have passed entrance examination conducted by the University.

### **Bachelor of Geomatic Engineering**

1. The Candidate who have scored 45% in aggregate in physical Group of 10+2 or minimum GPA 2.00 or intermediate level examination or I.E. Civil/Architect/Electronics/Computer Science or Diploma (minimum 2 years) in Geomatic/Civil Engineering from any Board or recognized by Purbanchal Universities.
2. The Candidate must have passed entrance examination conducted by the University.

### **Bachelor of Veterinary Science and Animal Husbandry**

1. The Candidate age is minimum 17 years complete.
2. The Candidate must have scored minimum C+ or GPA 2.00 in SLC or SEE.
3. The Candidate must have passed I. Sc. Examination or 10+2 (Science) with Physics, Chemistry, Biology & English from recognized Universities minimum 50% or GPA 2.40.
4. The Candidate must have passed entrance examination with minimum 50% marks conducted by the University.

*; dx -u\_ sf/nflm*

### **Bachelor of Computer Application (BCA)**

1. Should have successfully completed twelve years of schooling in any stream.
2. Minimum D+ grade in each subject of grade 11 and 12 with CGPA 2.0 or more

OR

Minimum score of second division (45%) marks in 10+2, PCL or equivalent in any discipline.

Students who have passed grade 11 and are waiting for supplementary exam (PURAK PARIKSHA) of grade 12 can also apply. However, they have to submit all the required documents at the time of admission.

Students who appeared in the final exam and are waiting for the result and certificates can also apply for the entrance examination. However, they have to submit all the required documents at the time of admission.

3. In case of foreign certificate, student should submit equivalence certificate and each subject grading with CGPA or total percentage document from concerned authority.
4. Should pass entrance examination as conducted by Purbanchal University.

### **Bachelor of Information Technology (BIT)**

1. Should have successfully completed twelve years of schooling in any stream with at least a mathematics subject of 100 marks in 10+2 level examination.
2. Minimum C grade in each subject of grade 11 and 12 with CGPA 2.0 or more. In case a student has taken the optional mathematics subject in grade 12, the student must have passed in the optional mathematics paper with at least D+ grade.

Or

Minimum score of second division (45%) marks in 10+2, PCL or equivalent in any discipline.

Students who have passed grade 11 and are waiting for supplementary exam (PURAK PARIKSHA) of grade 12 can also apply. However, they have to submit all the required documents at the time of admission.

Students who appeared in the final exam and are waiting for the result and certificates can also apply for the entrance examination. However, they have to submit all the required documents at the time of admission.

3. In case of foreign certificate, student should submit equivalence and each subject grading with CGPA or total percentage document from concerned authority.
4. Should pass entrance examination as conducted by Purbanchal University.

*;~~dx -3\_ sf]nflm~~*

### **Postgraduate Diploma in Computer Application (PGDCA)**

1. Students wishing to apply for the course must have completed graduation with a minimum of 10+2+3 years of typical education in any stream.
2. Minimum CGPA 2.0 or 45% marks.
3. Students who are waiting for final Exam result and certificates can also apply. However, they have to submit all the required documents at the time of admission.
4. In case of foreign certificate, student should submit equivalence certificate and each subject grading with CGPA or total percentage document from concerned authority.
5. Should pass entrance examination as conducted by Purbanchal University.

*:gftsf@/ txsf]z]l/s slo\$psf nflu*

### **Master of Computer Application (MCA)**

1. A Bachelor degree related to Computer Application, Information System, Information Technology, Information Science, Computer Science, Computer Engineering , Electronics and Communication Engineering, such as BSc Computer Science, BE Computer, BE Electronics and Communication, BScCSIT, BSc IT, BEIT, BE Software, BCA, BIT, PGDCA, BIM, BCIS, BBIS or equivalent.
2. Minimum CGPA 2.0 or 50% marks.  
Students who appeared in the final exam and are waiting for the result and certificates, can also apply for the entrance examination. However, they have to submit all the required documents at the time of admission.

3. In case of foreign certificate, student should submit equivalence certificate and each subject grading with CGPA or total percentage document from concerned authority.
4. Should pass entrance examination as conducted by Purbanchal University.

### **Master of Engineering in Earthquake (MEE)**

1. :gfts txsf] \$ j if[ Bachelor in Civil Engineering lj ifodf slDtdf Gogtd %) % j f Grading System sf] xsdf C+ (CGPA 2.4) kkt ul/ ptlOf{u/ \$f] xg' kg\$ .
2. In case of foreign certificate, student should submit equivalence certificate and each subject grading with CGPA or total percentage document from concerned authority.
3. Should pass entrance examination as conducted by Purbanchal University.

### **Master of Science in Information System Engineering**

1. :gfts txsf] \$ j if[ Bachelor in Computer Engineering ljifo / Bachelor Electronic & Communication Engineering j f ; f] ; /x lj ifodf slDtdf Gogtd %) % j f Grading System sf] xsdf C+ (CGPA 2.4) kkt ul/ ptlOf{u/ \$f] xg' kg\$ .
2. In case of foreign certificate, student should submit equivalence certificate and each subject grading with CGPA or total percentage document from concerned authority.
3. Should pass entrance examination as conducted by Purbanchal University.

### **Master of Science in Engineering Management**

1. :gfts txsf] \$ j if[ sg} klg OlGhlgol/8= lj ifodf slDtdf Gogtd %) % j f Grading System sf] xsdf C+ (CGPA 2.4) kkt ul/ ptlOf{u/ \$f] xg' kg\$ .
2. In case of foreign certificate, student should submit equivalence certificate and each subject grading with CGPA or total percentage document from concerned authority.
3. Should pass entrance examination as conducted by Purbanchal University.

### **MSc.in Nutrition and Dietetics**

1. :gfts txsf] \$ j if[ sg} klg OlGhlgol/8= lj ifodf slDtdf Gogtd %) % j f Grading System sf] xsdf C+ (CGPA 2.4) kkt ul/ ptlOf{u/ \$f] xg' kg\$ .
2. In case of foreign certificate, student should submit equivalence certificate and each subject grading with CGPA or total percentage document from concerned authority.
3. Should pass entrance examination as conducted by Purbanchal University.

### **MSc.in Urban Design and Conservation**

1. :gfts txsf] \$ j if[ / % j if[ Bachelor degree in Architecture, Civil engineering or other relevant fields df slDtdf Gogtd %) % j f Grading System sf] xsdf C+ (CGPA 2.4) kkt ul/ ptlOf{u/ \$f] xg' kg\$ .

2. In case of foreign certificate, student should submit equivalence certificate and each subject grading with CGPA or total percentage document from concerned authority.
3. Should pass entrance examination as conducted by Purbanchal University.

### **Master in Food Technology**

1. :gfts txsf] \$ j if] Bachelor degree in Science & Technology based curriculum such as B. Tech Food and B.Tech Dairy, or equivalent degree from recognized University af6 sldtdf Gogtd %) % j f Grading System sf] xsdf C+ (CGPA 2.4) kkt ul/ ptlof{u/\$f]xb' kg\$ .
2. In case of foreign certificate, student should submit equivalence certificate and each subject grading with CGPA or total percentage document from concerned authority.
3. Should pass entrance examination as conducted by Purbanchal University.

### **Master of Science in Agriculture (Agri-Business Management)**

1. :gfts txsf] B.Tech. or B.Sc. (Food/Dairy/Bio-Technology, Biochemistry), B.Sc.(Ag), or B.E.(Ag), B.V.Sc. & A.H., B.Sc.(Biological Science) or equivalent degree from recognized University af6 sldtdf Gogtd %) % j f Grading System sf] xsdf C+ (CGPA 2.4) kkt ul/ ptlof{u/\$f]xb' kg\$ .
2. In case of foreign certificate, student should submit equivalence certificate and each subject grading with CGPA or total percentage document from concerned authority.
3. Should pass entrance examination as conducted by Purbanchal University.

### **Master's of Science in Dairy Technology(M.Sc. Dairy Tech.)**

and

### **Master's of Science in Meat Technology(M.Sc. Meat Tech.)**

1. :gfts txsf] B.Tech. or B.Sc. (Food/Dairy/Bio-Technology) or equivalent, B.Sc or B.E. Agriculture, B.V.Sc. & A.H., B.Sc.Biochemistry and B.Sc. Microbiology or equivalent degree from recognized University af6 sldtdf Gogtd %) % j f Grading System sf] xsdf C+ (CGPA 2.4) kkt ul/ ptlof{u/\$f]xb' kg\$ .
2. In case of foreign certificate, student should submit equivalence certificate and each subject grading with CGPA or total percentage document from concerned authority.
3. Should pass entrance examination as conducted by Purbanchal University.

## Master of Information Technology(MIT)

1. Must have completed Bachelor degree related to Information Systems, Information Technology, Computer Application, Business Information Systems, Computer Information Systems, Information Science, Information Management, Information and Communication Technology, Computer Science, Computer Engineering , Electronics and Communication Engineering, such as BSc Computer Science, BE Computer, BE Electronics and Communication, BScCSIT, BSc IT, BEIT, BE Software, BCA, BIT, PGDCA,BICT, BIM, BCIS, BBIS or equivalent. Must have secured a minimum CGPA 2.0 or 50% marks. Students who appeared in the final examination of Bachelor degree and are waiting for the result and certificates, can also apply for the entrance examination. However, they have to submit all the required documents at the time of admission.
2. In case of foreign certificate, student should submit equivalence certificate and each subject grading with CGPA or total percentage document from concerned authority.
3. Should pass entrance examination as conducted by Purbanchal University.

### **cg, IT @ PURBANCHAL UNIVERSITY SCIENCE & TECHNOLOGY**

Name of College:

Address:

Contact Number:

Email:

Program:

Academic year:

SN	Name of the Student	Sex	Date of Birth	Permanent Address (VDC/Municipality/District)	PCL Percentage	Contact Number	E-mail Id	Remarks
1								
2								
3								

Approved By:

Name:

Designation: Campus Chief

Signature:

Date:

**College Seal**

**cg, PL #**  
**PURBANCHAL UNIVERSITY**  
**FACULTY OF SCIENCE AND TECHNOLOGY**

**Syllabus For Entrance Exam (Bachelor's Level)**

**Subject: English**

**F.M.: 15**

**Unit wise weightage**

<b>Units</b>	<b>Course content</b>	<b>Full marks</b>
A	Phonological questions, dealing with phonemes, Phonemic symbols and word stress	3
B	Laical Questions, Dealing with word formations and antonyms, a synonyms and one words for many words	2
C	Grammatical Question, dealing with the rest of the items given in the syllabus	10

**Unit A:**

Phonological questions, dealing with phonemes, Phonemic symbols and word stress

**Unit B:**

Laical Questions, Dealing with word formations and antonyms, a synonyms and one words for many words.

**Unit C:**

Grammatical Question, dealing with the rest of the items given in the syllabus.

Each Question is of objective type with multiple choice answer which carries one marks.

**Contents**

1. Use of Pronouns
2. Relative Pronouns
3. Sound
4. Stress
5. Intonation
6. Subject – Verb Agreements
7. Voice
8. Narration
9. Use of tense
10. Punctuation
11. Parts of Speech
12. Causative verb

13. Linking Verbs
14. Use of Prepositions
15. Sentence Structure
16. Transformation of Sentences
17. Conditional Sentences
18. To infinitive and gerund forms
19. Question Tag
20. Synonyms and Antonyms
21. One Word Substitution
22. Use of Suffix and Prefix
23. Word Power
24. The expression "ASIF" and "As Though"
25. The expression "NEED" and "In Need of"
26. Idioms and Phrases
27. Sentences: Simple, Compound and Complex
28. Affirmative and Negative Agreements
29. Use of Articles
30. Analogies

## Subject: Mathematics

F.M.: 35

### Unit wise weightage

Units	Course content	Full marks
1	Set and functions	5
2	Algebra	5
3	Trigonometry	10
4	Coordinate geometry	5
5	Calculus	5
6	Vectors	5

#### 1. Set and functions

Set, type of sets, operation on sets, law of sets, real number, Cartesian product, relations, functions and graphs, algebraic, exponential, trigonometric and logarithmic, hyperbolic functions and their inverse, basic properties of logarithmic functions.

#### 2. Algebra

Determinants and its properties, matrices, type of matrices, minor, cofactors and inverse of a matrix, uses of complex numbers, polynomial equation, sequences and series, permutation and combination, binomial theorem, exponential and logarithmic series.

### 3. Trigonometry

Trigonometric equations and general values, inverse trigonometric relations, inverse circular functions, principal values, properties of triangles, centroid, incentre, orthocenter and circumcentre and their properties.

### 4. Coordinate geometry

Coordinates in a plane, straight lines, pair of lines, circles, conic sections: parabola, hyperbola and ellipse, standard equations and simple properties, coordinates in space, plane and its equation.

### 5. Calculus

Limit and continuity, derivative and its application, rules of derivative, rate of change, maxima and minima of a function, integration, rules of integration, standard integral, definite integral and its application.

### 6. Vectors

Vectors in space, addition, subtraction and multiplication of vectors, unit vectors, linear combination of vectors, scalar and vector product of two vectors, application of vectors.

## Subject: Physics

F.M.: 30

### Unit wise weightage

Units	Course content	Full marks
1	Mechanics	8
2	Heat	4
3	Optics	4
4	Sound	4
5	Electricity	6
6	Atomic Physics	4

#### 1. Mechanics:

- Dimensions and Error analysis.
- Equations of motion.
- Motion of a projectile.
- Addition, Subtraction and multiplication of vectors, Resolution of a vector.
- Relative velocity.
- Laws of motion, Principle of conservation of linear momentum, Impulse, solid friction.
- Principle of moment, centre of mass, centre of gravity.
- Work, power and energy, principle of conservation of energy.
- Centripetal force and its application.
- Moment of inertia, Torque on a body, angular momentum and its conservation, work done by couple, kinetic energy of rolling body.

- Law of gravitation, Gravitational potential, Gravitational field intensity, Escape velocity, Total energy of a satellite, Kepler's law of planetary motion.
- Characteristics of S.H.M., Energy of a particle exciting S.H.M., Simple pendulum.
- Elasticity, Stress & Strain, Modulus of elasticity, Energy Stored in a stretched wire.
- Surface tension of liquid, Surface energy, Capillarity.
- Viscosity of fluid, coefficient of viscosity, Stoke's law.
- Terminal velocity, Energy of fluid.

## 2. Heat:

- Heat and Temperature, Scales of temperature, Thermal equilibrium.
- Measurement of heat, Specific heat capacity, thermal capacity, Latent heat.
- Expansion of Solid, liquid and gas.
- Gas laws, Ideal gas equation.
- Kinetic theory of gas. Root mean square speed of gas molecules.
- Transfer of heat, Conduction, Convection and Radiation, Stefan's law, Kirchhoff's law.
- Relative humidity and dew point.
- First law of thermodynamics, Isothermal and adiabatic processes.
- Second law of thermodynamics, Carnot's engine, Entropy.

## 3. Optics:

- Formation of images by plane and curved mirrors.
- Refraction of light through plane surface, Refractive index, Critical angle, Total internal reflection.
- Refraction through lenses, Achromatic combination of two lenses.
- Visual angle, Angular magnification, Microscope and Telescope.
- Interference, Diffraction and Polarization of light.

## 4. Sound:

- Damped and forced oscillation, Resonance, progressive waves, principle of superposition.
- Velocity of Sound in solid, liquid and gas, Laplace's correction.
- Beat phenomena.
- Doppler effect.
- Stationary waves, waves in pipe, waves in string.

## 5. Electricity:

- Electric charge, Electrostatic induction, Surface charge density.
- Electric field, Electric potential, Electric field intensity, Gauss's law and its applications.
- Capacitors, Dielectric Strength.
- Metallic Conduction, Ohm's law, Resistance, Conductance, Resistivity, Conductivity, Combination of resistance.
- Emf, Potential difference, internal resistance of a cell, Combination of cells.
- Heating effect of Current, Joule's law, Electric power.

- Kirchhoff's law and its application.
- Galvanometer, Conversion of galvanometer into voltmeter and ammeter.
- Earth's magnetism.
- Magnetic field, Magnetic flux, Force on Current Carrying Conductor, Biot-Savart's law and their applications. Ampere's law.
- Electromagnetic induction, Faraday's law and Lenz's law, Emf in rotating coil.
- AC circuits.

**6. Atomic Physics:**

- Discharge of electricity through gases, Cathode rays, Electronic mass and charge.
- Bohr's theory of atomic model, Energy level diagram.
- X-rays, Photoelectric effect.
- Radioactivity, Decay law, Half-life period.
- Nuclear fission and fusion.
- Semiconductors, junction diode.

**Subject : Chemistry**

F.M. : 20

**Unit wise weightage**

Units	Course content	Full marks
1	STATES OF MATTER	2
2	ATOMIC STRUCTURE	3
3	OXIDATION AND REDUCTION	2
4	PERIODIC TABLE	2
5	HYDROGEN, OXIGEN, OZONE AND WATER	3
6	INTRODUCTION TO ORGANIC CHEMISTRY AND HYDROCARBON	2
7	VOLUMETRIC ANALYSIS AND IONIC EQUILIBRIUM	2
8	ALDEHYDES AND KETONES	2
9	IRON	2

**1. STATES OF MATTER:**

- The gas laws (Boyle's law Charle-law,combined gas law)
- Kinetic molecular theory of gases
- Ideal and real gases.
- Vander Waals Equation.
- Properties of liquid
- Solutions
- Concentration of solutions
- Saturated, unsaturated and super saturated solutions.
- Solubility and its determination
- Efflorescence and Deliquescence
- Water of crystallization

## 2. ATOMIC STRUCTURE

- Fundamentals particles and their discovery.
- Thomson's Atomic Model
- Rutherford's Atomic model
- Atomic number and Mass number.
- Composition of an atom.
- Isotopes and isobars.
- Bohr's atomic model
- de Broglie's Relation.
- Heisenberg's uncertainty principle.
- Quantum mechanical model of atom.
- Atomic orbitals.
- Quantum number.
- Pauli's Exclusion principle.
- Hund's rule of maximum multiplicity.
- Aufbau principle and electronic configuration of elements.

## 3. OXIDATION AND REDUCTION

- Classical concept of oxidation .
- Electronic concept of oxidation and Reduction.
- Oxidising and Reducing agents.
- Oxidation number
- Redox reactions and Electrolysis.

## 4. PERIODIC TABLE

- Mendeleev's periodic Table
- Advantage and Defect of Mendeleev's Periodic Table.
- Modern periodic law.
- Long form of periodic Table.
- Features of Long form of periodic Table
- Defects of Long form of periodic Table.
- Grouping of Elements into Blocks.
- Bohr's classification of elements .
- Periodic properties and their periodic trends.
- Atomic radius, Ionic radius, Ionization energy,
- Electron affinity and Electronegativity

## 5. HYDROGEN, OXYGEN, OZONE AND WATER

- Position of Hydrogen in the periodic table.
- Preparation and properties of hydrogen.
- Isotopes of hydrogen
- Position of oxygen in periodic table
- Preparation and properties of oxygen.
- Oxides.
- Structure of Ozone
- Ozone layers and its depletion.
- Composition of water and structure of water.
- Properties of water
- hard and soft water and its removal.
- Heavy water.
- Properties and uses of heavy water.

## 6. INTRODUCTION TO ORGANIC CHEMISTRY AND HYDROCARBON

- Definition of organic Chemistry and organic compound.
- vital force theory and its limitations
- Tetra valency and catenation property of carbon.
- Functional group and homologous, series
- Meaning of empirical, molecular, structural and contracted formula.
- Qualitative analysis of organic compounds
- IUPAC naming of organic compounds
- Structural isomerism and its type.
- Concept of homolytic fission Heterohytic fission electrophiles nuleophiles and inductive effect
- Preparation of alkynes and its properties with  $H_2, X_2, HX, H_2O, O_3, H_2SO_4$  Baeyer's reaction.
- Preparation of alkynes and its properties with  $H_2, X_2, HX, H_2O, O_3$  Acidic nature, Action with ammonical  $AgNO_3$  with alkaline  $kinno_4$  and polymerization reaction.

## 7. VOLUMETRIC ANALYSIS AND IONIC EQUILIBRIUM

- Acidimetry and aclkalimetry
- Equivalent mass of compounds
- Expressing concentration interms of Normality, Morality and Molality
- Principles of volumetric analysis
- Theory of chemical indicators and selection of an indicators
- Classification of Electrolytes
- Arrhenius Theory of Ionization
- Ionisation of water, solubility product and Communion effect
- Arrhernius concept of acid and base
- Bronsted concept of acid and base

## 8. ALDEHYDES AND KETONES

- Preparation of aldehydes and ketones from dehydrogenation and oxidation of alchohol, ozonolysis of alkene, calalytic hydration of alkynes
- Physical properties of aldelydes and Ketones.
- Chemical properties –Addition reaction, reaction with  $H_2, HCN, NaHSO_3$ , Grignard reagents  $NH_2-NH_2$ , Phenyl-hydrazine, semicarbazide and 2,4-DNP.
- Reduction properties of aldehyde-oxidation with Tollen's reagent, Fehling solution
- Aldol condensation clemennson's reduction Wolf-Kishner reduction Action with  $PCl_5$
- Preparation of benzaldehyde from Toluene
- Chemical Properties-Perkin condensation, Bcenzoin condensation, Cannizzaro's reaction

## 9. IRON

- Occurrence and extraction
- Varities of iron preparation of iron.
- Manufacture of steel by-Bessemer process and open hearth process
- Heat treatment of Steel.
- Stainless steel.
- Rusting of iron and its prevention
- Biological importance of iron
- Structure and uses of green vitrol, Ferric Chloride, Mohr's saclt.

Unit wise weightage

Units	Course content	Full marks
1	Bio-molecules and cell Biology	3
2	Biodiversity	8
3	Biota and their Environment	4
4	Anatomy and Physiology of organisms	3
5	Genetics	4
6	Developmental Biology	3
7	Human Biology and Health	5
8	Application of Biology	5

**Unit 1: Bio-molecules and cell Biology**

**Bio-molecules:** Carbohydrate, protein, lipid, nucleic acid and minerals.

**Cell Biology:** Prokaryotic and eukaryotic cells, cell organelles, cell division (amitosis, mitosis and meiosis).

**Unit 2: Biodiversity**

Concepts on taxonomy (definition, nomenclature and classification)

**Monera:** Bacteria and *Nostoc*.

**Mycota:** General characters of Zygomycetes, Ascomycetes, Basidiomycetes and deuteromycetes; economic importance of fungi; morphology and reproduction of Mucor and Yeast.

**Plantae:** General characters of algae (green, brown and red), bryophyta, pteridophyta, gymnosperm and angiosperm; morphology and reproduction of Spirogyra, Marchantia, Funaria, Pteridium, Cycas and Pinus; distribution of Pinus in Nepal; morphology and taxonomy of Brassicaceae, Asteraceae, Solanaceae, Papilionaceae and Poaceae; structure and economic importance of Lichens and Virus.

**Protista:** General characters and classification of Protozoa; habit, habitat, structure and reproduction of Paramecium and Plasmodium.

**Animalia:** General characters and classification (upto class) of Porifera, Coelenterata (Cnidaria), Platyhelminthes, Aschelminthes (Nema-theminthes), Annelida, Arthropoda, Mollusca, Echinodermata and Chordata; Earthworm (structure; digestive, circulatory, excretory, reproductive and nervous systems); Frog (structure; digestive, circulatory, respiratory, urino-genital, reproductive and nervous systems).

**Unit 3: Biota and their Environment**

Ecology (definition, ecological factors and their interactions); structure and functional aspects (food chain and food web, trophic level, ecological pyramids) of pond and grassland ecosystems; concepts of community and successions; nitrogen cycle, acid rain, green house effects, depletion of ozone layer; forests of Nepal; meaning of rare, threatened, vulnerable and endangered species; national parks, wildlife reserves and hunting reserves of Nepal.

**Unit 4: Anatomy and Physiology of organisms**

Concepts on plant and animal tissues, internal structure of dicot and monocot stems and roots, secondary growth on dicot stem; osmosis, diffusion, water absorption, transpiration, photosynthesis, respiration, growth hormones (auxins, gibberellins, cytokinins), concept on plant movement.

### Unit 5: Genetics

Genetic materials: DNA (composition, structure and replication), RNA (types with functions); genetic code, Mendel's laws of inheritance, concept of incomplete dominance and co-dominance, epistasis, polygenic inheritance (skin colour in men), pleiotropic gene, linkage and crossing over, sex linked inheritance (colour blindness in man), mutation and polyploidy.

### Unit 6: Developmental Biology

Angiosperms (asexual reproduction, development of male and female gametophytes, pollination and fertilization)

### Unit 7: Human Biology and Health

Nutrition, digestion, respiration, circulation (blood groups and Rh-factor, heart structure and action), arterial and venous systems, excretion and osmoregulation; functions of endocrine glands (pituitary, thyroid, parathyroid, pancreatic and adrenal glands); reproduction; structure and functions of eye and ear; human disease (drug abuse, alcoholism, smoking, typhoid, AIDS, cancer)

### Unit 8: Application of Biology

Biotechnology (branches, applications in agriculture, medicine, industries, conservation, etc.), tissue culture (types, methods: sterilization, composition of medium and its preparation); plant breeding; green manure; concept on antibiotics and vaccines; amniocentesis and test tube baby; genetic engineering (definition, tools, steps, applications); methods and applications of alcohol and antibiotics fermentation.

## Subject: Computer Fundamental

F.M.: 30

### Unit wise weightage

Units	Course content	Full marks
1	Introduction to Computer	3
2	Basic Computer Organization and Computer Peripherals	4
3	Computer Storage	2
4	Computer Software	2
5	Introduction to Database	2
6	Networks and Internet	4
7	Computer Hardware	2
8	Basics of Windows and User Interface	3
9	Basic DOS Commands	2
10	Word Processing	2
11	Spreadsheets	2
12	Presentations	2

### Unit 1: Introduction to Computer:

- Characteristics, applications, and components of computer
- Classification of computer based on purpose, size and technology

**Unit 2: Basic Computer Organization and Computer Peripherals:**

- Block diagram of computer system
- Input devices: Keyboard, mouse, and other types of input devices
- Output devices: Monitor, printer, and other types of output devices

**Unit 3: Computer Storage:**

- Concepts of memory and requirements of storage devices
- Classification and types of storage devices
- ROM and RAM with their types
- Magnetic devices and Optical devices

**Unit 4: Computer Software:**

- Introduction and types of software
- Definition and functions of operating system
- Programming languages and their types

**Unit 5: Introduction to Database:**

- Meaning of data and information
- Concepts and characteristics of database and DBMS

**Unit 6: Networks and Internet:**

- Introduction and uses of network
- Types and topologies of network
- Introduction, features and applications of Internet
- Concepts of intranet and extranet
- Network media and network software
- WWW, E-mail, E-commerce, web site
- Web browsing, net surfing, chatting, using e-mails
- Computer crime, viruses and threats
- Cyber law and ethical issues

**Unit 7: Computer Hardware:**

- Motherboard and its parts, slots, daughterboard, and expansion slots
- BIOS, CMOS, and Microprocessor
- utilities and application software, Customizing software

**Unit 8: Basics of Windows and User Interface:**

- Using mouse and moving icons on the screen
- The My Computer icon, the Recycle Bin icon, Status Bar, Start button, Menu Bar
- Opening, closing and running an application
- Using Windows Explorer to view files, folders and directories
- Creating and renaming files and folders
- Windows settings: control panel, wallpapers, screensavers, date and time, sound
- Advanced features: using right mouse button, shortcuts, notepad, accessories

**Unit 9: Basic DOS Commands:**

- Comparison of DOS and Windows, External and Internal Commands.

**Unit 10: Word Processing:**

- Basics: opening and closing documents, saving documents, page setup, printing, scrolling around a document

- Text manipulation and formatting: text selection, cut, copy and paste, font, Bold, Italic and Underline, text alignment, line and paragraph setting, changing font, size and color, bullets and numbering, changing case
- Table manipulation: drawing and inserting table, changing cell width and height, alignment of text in cell, inserting and deleting rows and columns, table borders

**Unit 11: Spreadsheets:**

- Basics: opening and closing of spreadsheet, multiple sheets, Menu Bar, cell inputting, cell addressing
- Manipulation of cells: entering texts, creating tables, setting cell width and height, copying of cells
- Formulas: sum, average, percentage, and other basic functions
- Preparing invoices/budgets, totaling of transactions, maintaining daily and monthly reports

**Unit 12: Presentations:**

- Basics: opening a PowerPoint presentation, using Wizard to create a presentation
- Slide presentation: title, text, picture, table, font color and font size, bullets and indenting, slide design, background, slide numbering, slide show, slide animation, slide sorting, printing slides

**Subject: Aptitude**

**F.M.: 30**

S.No.	Topics	Marks
1	Sentence Completion	2
2	Analogies	3
3	Word Groups	2
4	Logical Reasoning	3
5	Deduction	2
6	Numerical Computation	2
7	Numerical Estimation	2
8	Numerical Reasoning	3
9	Percentage, Ratio and Proportion	3
10	Data Interpretation	2
11	History of Nepal	2
12	Geography of Nepal	2
13	General Information of PU	2

## Model questions (Biology)

- Which cell organelle is called suicidal bag of the cell?  
a. Mitochondria      b. Chloroplast      c. Lysosome      d. Ribosome
- Coenocytic mycelium is found in  
a. Mucor      b. Yeast      c. Spirogyra      d. Alternaria
- Cruciform corolla is found in  
a. Mustard      b. Pea      c. Sunflower      d. Maize
- Which of the following structure helps in reproduction in Paramecium?  
a. Macronucleus      b. Micronucleus      c. Food vacuole      d. Contractile vacuole
- In which of the segment female genital aperture in earthworm is situated?  
a. 14      b. 15      c. 17      d. 18
- Pyramid of biomass in pond ecosystem is  
a. Upright      b. Inverted      c. Both a and b      d. None
- The largest national park of Nepal is  
a. Chitwan      b. Shey Phoksundo      c. Bardia      d. Langtang
- Which one of the following components of xylem is living?  
a. Tracheid      b. Vessel      c. Xylem parenchyma      d. Xylem fibre
- ABO blood group system in human is an example of  
a. Incomplete dominance      b. Co-dominance      c. Epistasis      d. Linkage
- How many sex chromosomes are present in a human cell?  
a. 46      b. 44      c. 22      d. 2
- How many meiotic divisions are required to produce 100 seeds (zygotes)?  
a. 125      b. 100      c. 50      d. 25
- Name the disease caused due to deficiency of iodine  
a. Hyponatremia      b. Goitre      c. Rickets      d. Osteomalacia
- How many pairs of cranial nerves are present in man?  
a. 12      b. 31      c. 40      d. 10
- Which one of the following plant is extensively used as green manure in Nepal?  
a. *Crotalaria juncea*      b. *Brassica campestris*  
c. *Solanum nigrum*      d. *Zea mays*
- The vectors used in genetic engineering are  
a. Plasmids      b. Cosmids      c. Bacteriophages      d. All

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